THE SYNTHESIS OF NAPHTH[2,1-a]- AND NAPHTH[2,3-a]AZULENES1)

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Naphth[2,1-a]- and naphth[2,3-a]azulenes were synthesized by dehydrogenation of 5,6-dihydronaphth[2,1-a]- and 5,12-dihydronaphth[2,3-a]azulenes which were obtained by the reaction of 2H-cyclohepta[b]furan-2-one with 1- and 2-pyrrolidiny1-3,4-dihydronaphthalenes, respectively.

The polycyclic aromatic hydrocarbons which consist of the azulene ring condensed with benzenoid aromatics are interest in their physical properties and physiological activities,  $^{2}$  as well as the chemical behaviors. Among of these compounds, naphth[a]azulenes, such as naphth[2,1-a]-  $(\underline{1})$  and naphth[2,3-a]azulenes  $(\underline{2})$ , are fundamental compounds as higher benzologs of benz[a]azulene,  $^{3}$  but any

isomer of naphth[a]azulene has not yet been synthesized. The present authors have previously reported the useful method for synthesizing the azulene ring utilizing the reaction of 2H-cyclohepta[b]furan-2-one ( $\underline{3}$ ) with enamines<sup>4</sup>) and the reaction has been applied on the facile synthesis of indenoazulenes.<sup>5</sup>) This communication describes the synthesis of  $\underline{1}$  and  $\underline{2}$  from  $\underline{3}$  via dihydronaphth[a]azulenes which were obtained by the reaction of  $\underline{3}$  with the enamines of  $\alpha$ - and  $\beta$ -tetralones, that is, 1- ( $\underline{4a}$ ,  $\underline{b}$ ,  $\underline{c}$ )<sup>6</sup>) and 2-pyrrolidiny1-3,4-dihydronaphthalenes ( $\underline{5}$ ).<sup>7</sup>)

The Synthesis of Naphth[2,1-a]azulene ( $\underline{1a}$ ) and Its 2-Methyl and 2-t-Butyl Derivatives ( $\underline{1b}$  and  $\underline{1c}$ ). A solution of 3 and 2 molar equivalents of  $\underline{4a}$  in

anhydrous ethanol was heated under reflux for 4 hr. At that time, one more molar equivalent of 4a was added and the heating was continued for an additional 4 hr. After evaporation of the solvent, the residue was chromatographed (silica gel, benzene-cyclohexane). The bluish fraction afforded 5,6-dihydronaphth[2,1-a]azulene (6a) [blue scales, mp 169-170°C] ) in a 50% yield. In a similar manner, the reaction of 3 with 7-methyl (4b) and 7-t-butyl derivatives (4c) of 4a yielded 2-methyl (6b) [blue scales, mp 156-157°C] and 2-t-butyl derivatives (6c)[bluish green needles, mp 142°C] of 6a in 45 and 38% yields, respectively. On heating in p-cymene in the presence of 5% Pd-C, 6a, b, c resulted in dehydrogenation to give la [green scales, mp 253-254.5°C], lb [green scales, mp 200-201°C] and lc [green scales, mp 180-181°C] in 52, 50, and 55% yields, respectively. The dehydroqenation of 6a, b, c with one molar equivalent of DDQ in benzene also yielded la, b, and c in 68, 65, and 68% yields, respectively. Further, the dehydrogenation of 6a into la was achieved in a satisfactory yield as follows: A trifluoroacetyl derivative (7) [brown oil], derived from 6a by treatment with (CF3CO)2O, was hydrolyzed with aq. NaOH in ethanol, and the subsequent methylation of the resulting carboxylic acid with diazomethane gave an ester (8) [violet oil] in an 85% yield from 6a. The dehydrogenation of 8 with DDQ in benzene yielded methyl naphth[2,1-a]azulene-12-carboxylate (9) [green needles, mp 120-121°C] in an almostquantitative yield. On heating with 100% phosphoric acid,  $\frac{9}{2}$  resulted in demethoxycarbonylation to give la in an almost quantitative yield. The spectral data of la, b, and c were shown in Table 1.

$$\frac{10: X=H}{12: X=COCF_3}$$

$$\frac{10: X=H}{12: X=COCF_3}$$

$$\frac{11: X=H}{13: X=COCF_3}$$

$$12$$

$$\frac{12: X=COCF_3}{13: X=COCF_3}$$

The Synthesis of Naphth[2,3-a]azulene (2). A solution of 3 and 3 molar equivalents of the enamine, 5, in anhydrous ethanol was heated under reflux for 462 hr. After evaporation of the solvent, the residue was chromatographed (silica gel, benzene-cyclohexane) and the bluish fraction was recrystallized from benzene to give 5,12-dihydronaphth[2,3-a]azulene ( $\underline{10}$ ) [blue scales, mp 207-209°C, pmr (CDCl<sub>3</sub>)  $\delta$  4.50 ppm (4H, s, CH<sub>2</sub>)] in a 20% yield. An expected 5,6-dihydronaphth[1,2-a]azulene (11) could not be isolated. However, the trifluoroacetylation of mother liquor of the recrystallization yielded 7-trifluoroacety1-5,6-dihydronaphth-[1,2-a]azulene ( $\underline{13}$ ) [brown oil, pmr (CDCl<sub>3</sub>):  $\delta$  3.33 (2H, t, J = 7.0 Hz, CH<sub>2</sub>) and 3.85 ppm (2H, t,  $J = 7.0 \, Hz$ ,  $CH_2$ )] as a minor product, as well as ll-trifluoroacety1-5,12-dihydronaphth[2,3-a]azulene ( $\frac{12}{2}$ ) [brown oil, pmr(CDCl<sub>3</sub>):  $\delta$  4.27 (2H, br. s,  $CH_2$ ) and 4.57 ppm (2H, br. s,  $CH_2$ )] which was derived from  $\underline{10}$  by treatment with  $(CF_3CO)_2O$ . The structures of 12 and 13 were established on the basis of their pmr spectral data. The enamine (5) is known to react with electrophiles not only at the 1-position, but also at the 3-position, in spite of the minor contribution of a tautomer  $(\underline{5a})$ . In the case of  $\underline{3}$ , the reaction took place more easily with 5a rather than 5, giving mainly 10.

The treatment of  $\underline{12}$  with DDQ resulted in an easy dehydrogenation to give lltrifluoroacetylnaphth[2,3-a]azulene ( $\underline{14}$ ) [brownish violet needles, mp 177.5-178°C] in a quantitative yield. The hydrolysis of  $\underline{14}$  with aq. NaOH in ethanol and the subsequent decarboxylation of the resulting carboxylic acid with 100% phosphoric

acid yielded  $\underline{2}$  [green scales, mp 223-224°C] in an almost quantitative yield. Naphth[1,2-a]azulene could not be synthesized because of a low yield of  $\underline{13}$ . The spectral data of 2 were shown in Table 1.

Table 1. The UV and IR spectral data of naphth[a]azulenes,  $\underline{la}$ ,  $\underline{lb}$ ,  $\underline{lc}$ ,  $\underline{9}$ , and  $\underline{2}$ .

	The state of the s	0		
	UV in MeOH, λmax nm	(log $\epsilon$ )	IR (KBr)	cm-1
	272 (4.24), 322 (4.87),	375 (3.92) 15	570, 1460, 1442,	1280, 1190
<u>la</u>	395 (3.54), 411 (3.17),	580 (2.46)	932, 800, 742,	721, 705
	630 (2.57)			
	270 (3.89), 316 (4.54),	326 (4.59) 15	570, 1530, 1475,	1380, 1290
<u>lb</u>	374 (3.89), 395 (3.91),	599 (2.00) 12	205, 938, 880,	750, 735
	638 (2.08)			
	270 (3.90), 315 (4.51),	327 (4.56) 15	592, 1578, 1470,	1373, 1285
<u>lc</u>	375 (3.90), 396 (3.93),	580 (2.46) 13	180, 1015, 935,	880, 830
	640 (2.55)			
	270 (4.21), 278 (4.21),	319 (4.20) 16	85, 1595, 1580,	1455, 1410
<u>9</u>	355 (4.24), 396 (3.91),	618 (2.57) 14	102, 1220, 1210,	1195, 1130
		13	100, 800, 750,	745
	269 (4.41), 279 (4.38),	316 (4.77) 16	520, 1586, 1490,	1466, 1402
<u>2</u>	326 (4.76), 366 (3.85),	392 (3.72) 13	392, 1330, 1304,	1262, 1238
	414 (3.66), 586 (2.52),	647 (2.60) 12	205, 1110, 1006,	946, 920
		8	378, 810, 723,	668

This research has been supported by a Grant-in-Aid for Scientific Research from the Japanese Ministry of Education.

## References and Notes

- 1) Part III of the series of the syntheses of azulene derivatives by the reactions of 2H-cyclohepta[b]furan-2-one with enamines. Part II see ref. 5).
- a) D. J. Bertelli and P. Crews, Tetrahedron, <u>26</u>, 4717 (1970), b) C. Jutz,
   E. Schweger, Hans-Goerg Lovering, A. Kraatz, and W. Kosban, Chem. Ber., <u>107</u>,
   2956 (1974), c) N. P. Buu-Hoi, D. P. Hien, C. Jutz, Naturwissenschaften, <u>54</u>,
   420 (1967); N. P. Buu-Hoi, N. B. Ciao, C. Jutz, ibid., 57, 499 (1970).
- 3) Pl. A. Plattner, A. Furst, J. Chopin, and G. Winteler, Helv. Chem. Acta, 31, 501 (1948); W. Treibs, Chem. Ber., 81, 381 (1948).
- 4) P. W. Yang, M. Yasunami, and K. Takase, Tetrahedron Lett., 1971, 4725.
- 5) Alice Chen, M. Yasunami, and K. Takase, Tetrahedron Lett., 1974, 2581.
- 6) L. H. Hellberg, R. J. Milligan, and R. N. Wilke, J. Chem. Soc. (C) <u>1970</u>, 2581.
- 7) G. Pitacco, F. Paolo Colonna, E. Valentin, and A. Risaliti, J. Chem. Soc. Perkin I, 1974, 1625.
- 8) All new crystalline compounds gave satisfactory results of elementary analyses in accord with the assigned structures.